

Superconducting properties of the oxygen-deficient iron oxyarsenide $\text{TbFeAsO}_{1-\delta}$ from underdoped to overdoped compositions

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A wide-range doping was achieved by a high-pressure method for $\text{TbFeAsO}_{1-\delta}$ from “underdoped” to “overdoped” superconducting compositions throughout the optimized superconductivity (T_c of 44 K). T_c vs δ shows a dome-shaped feature, while T_c vs the lattice constant likely follows a unique empirical curve over the doping range. The relatively large amount of oxygen vacancies up to ~ 0.3 per the formula unit was introduced possibly because of the smaller replacement Tb than the other Ln (rare-earth element) in the $\text{LnFeAsO}_{1-\delta}$ system.

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I. INTRODUCTION

Discovery of high- T_c superconductivity in the iron oxyarsenide LnFeAsO probably caused one of the highest impacts in materials science since 1986, the year copper oxide superconductor was discovered.^{1,2} Reflecting the significant attention, large number of manuscripts has been submitted to the preprint archive about the superconductor since then.³ It is also noticeable that development of varieties of the iron superconductor advanced swiftly; for example, such as the “infinite” system AFe_2As_2 ,^{4,5} the iron phosphide oxide,^{6,7} and the iron selenide^{8,9} were discovered to be superconducting within several months right after the discovery. However, the earlier system LnFeAsO still holds the highest T_c record among them and keeps capturing scientific and practical attentions: the highest T_c to date is 56 K for $\text{SmFeAs}(\text{O},\text{F})$.¹⁰ Besides, the upper critical field is remarkably high,¹¹ suggesting many possibilities of future applications. Meanwhile, we probably have a broad consensus that crystal growth of the doped LnFeAsO is highly challenging. Due to lack of the high-quality crystals in experimental studies, a quality-improved and carefully controlled polycrystalline sample still holds high values for continuous progress of understandings of the iron superconductor.

The parent material LnFeAsO crystallizes in a layered structure (ZrCuSiAs-type),⁷ consisting of alternative stacking of Fe_2As_2 layer and Ln_2O_2 layer along a crystallographic direction. The Fe_2As_2 layer plays an essential role to establish the $3d$ multi bands which are directly responsible for the superconductivity, and the insulating Ln_2O_2 layer plays a role as a charge-carriers source by accommodating doped elements and oxygen vacancies.¹² There are many studies in progress, however nature of the superconductivity seems to be under debate: superconducting symmetry was suggested to be fully gaped, while a pseudogap with line node was suggested instead.¹³ A delicate balance between the degree of

antiferromagnetic $3d$ spin fluctuations and the pair-breaking ferromagnetic interactions is possibly significant in determination of T_c .¹² On the other side, superconducting properties are highly sensitive to local structure environments such as the coordination of iron by arsenic.¹⁴ Although systematic studies are achieved in a wide variety of ways to some extent, further studies seems be needed to elucidate the principal physics of the superconductivity in the LnFeAsO system.

In this study, we focused on $\text{TbFeAsO}_{1-\delta}$ in which oxygen vacancies are introduced to turn on the superconductivity without fluorine usually doped with. We found that T_c goes up and then goes down in a systematic way with increasing the oxygen vacancies concentration. Indeed, a bell-shaped feature in T_c vs δ was observed as well as what was found for the copper oxide superconductor. To our best knowledge, this would be the first observation of the bell-shaped feature in the LnFeAsO system.

II. EXPERIMENTAL

Details of the sample preparation for $\text{TbFeAsO}_{1-\delta}$ were reported elsewhere.¹⁵ A high-pressure synthesis method was used. We prepared samples at $\delta=0, 0.10, 0.15, 0.20,$ and 0.25 (the first group, notated as #1), and several months later we synthesized again at $\delta=0.10, 0.15, 0.18, 0.20,$ and 0.25 as the second group (#2) to test reproducibility of the initial synthesis. Afterwards, we synthesized additional samples at $\delta=0.25$ and 0.30 (#3). An allied sample $\text{LaFeAsO}_{0.85}$ was prepared for a comparison in the same manner. A La rod (99.9%, Nilaco Co.) was powdered by scratching the surfaces in a globe box, and was employed as a starting material instead of the Tb source.

The sample quality was investigated by a powder x-ray diffraction (XRD) method in a commercial apparatus (Rigaku, RINT2200V/PC) using $\text{Cu-K}\alpha$ radiation. Crystal structure of the selected samples $\text{LaFeAsO}_{0.85}$ and

TbFeAsO_{0.85} (#1) were studied by a synchrotron x-ray diffraction (SXRD) method at ambient temperature/pressure in a large Debye-Scherrer camera at the BL15XU beam line of SPring-8.¹⁶ Incident beam was monochromatized at $\lambda = 0.65297$ Å, and a sample capillary (Lindenmann glass made, outer diameter was 0.1 mm) was rotated during the intensity measurement to reduce the preferred-orientation effect. Oxygen content of the selected samples was measured by a gravimetric method.

Electrical resistivity (ρ) of the polycrystalline TbFeAsO_{1- δ} was measured by a four-point probe method with a gage current of 0.1 mA ($\delta=0.10$) and 1 mA (0.15, 0.18, 0.20, 0.25, 0.30) in a Quantum Design PPMS. Electrical contacts on the four locations along the rectangular parallel piped sample were prepared from gold wires and silver paste. The measurements were conducted between 2 and 300 K. The $\delta=0.15$ sample (#1) was selected to study a magnetic field dependence (≤ 90 kOe) of ρ . Magnetic susceptibility (χ) of the samples was measured in a Quantum Design MPMS. The sample each was cooled to 2 K without applying a magnetic field, and then the sample was slowly warmed up in a magnetic field of 10 Oe [zero-field cooling (ZFC)], followed by cooling down to 2 K [field cooling (FC)]. Isothermal magnetization of a selected sample was measured at 2 K up to 200 Oe in the MPMS. Specific heat (C_p) of the samples was measured in the PPMS by a time-relaxation method.

Ac magnetic-susceptibility $\chi' - i\chi''$ was measured on the polycrystalline TbFeAsO_{0.85} (#1) in a laboratory-made magnetometer, which comprises the Hartshorn bridge and a temperature controller. Balance of the electrical bridge was carefully maintained during the measurements. A two-phase lock-in amplifier was used to detect in-phase (χ') and out-of-phase (χ'') signals of the bridge over the superconducting transition. An ac magnetic field was applied to the sample during the measurements without magnetic shielding; the applied ac field was thus superimposed on the Earth's field of ~ 300 mOe. We used $H_{ac}=100$ mOe, where a fundamental frequency was 132 Hz. Details of the magnetometer were reported elsewhere.¹⁷

III. RESULTS AND DISCUSSION

Figure 1 shows XRD profiles of the polycrystalline TbFeAsO_{1- δ} at $\delta=0, 0.10, 0.15, 0.20,$ and 0.25 (#1) and $\delta=0.18$ (#2) and $\delta=0.30$ (#3). We carefully compared the XRD profiles of the samples having the same δ , and found no significant difference, suggesting high degree of synthesis reproducibility. Over the entire profiles, major peaks were clearly indexed by the hkl numbers as shown in Ref. 15, calculated from a tetragonal structure model, while the $\delta=0$ profile has nontrivial impurities peaks (marked by stars). It is notable that the low sample quality at $\delta=0$ was dramatically improved by introducing an amount of the oxygen vacancy. We found even 10 mol % of the oxygen vacancy, i.e., $\delta=0.10$, sufficiently improved the sample quality, suggesting two possibilities at least. (i) The nominal oxygen quantity was underestimated by approximately 0.1 mol per the formula unit (f.u.) and the oxygen stoichiometry was practically

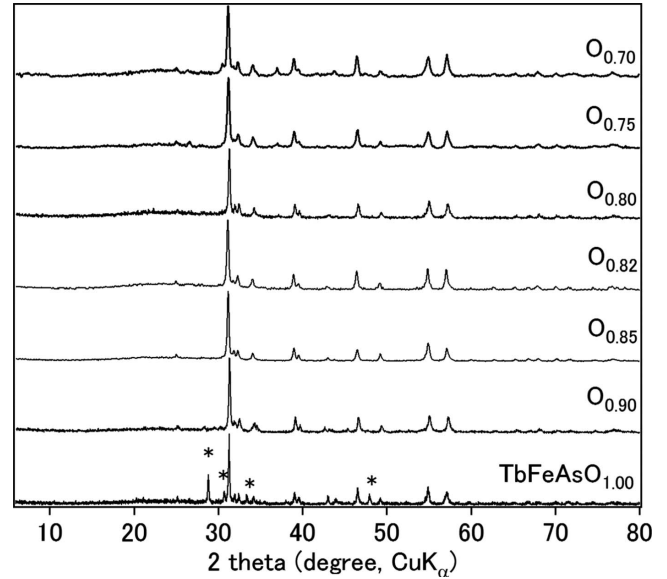


FIG. 1. XRD profiles of the polycrystalline TbFeAsO_{1- δ} . Nominal composition is indicated at the profile each. Peak identification was successful with the tetragonal structure model ($P4/nmm$) as was in Ref. 15. Stars indicate peaks due to impurities including Tb₂O₃.

realized at $\delta=0.1$. (ii) High-pressure chemical reaction is sluggish in the oxygen-defect free system. A useful hint was obtained by subsequent magnetic studies (the data are shown later). A superconducting transition was observed at $\delta=0.1$ and was not at all (>2 K) at $\delta=0$. Considering the lattice parameters which were certainly changed by introducing the oxygen vacancies (Fig. 2), it is thus likely that the latter is primarily responsible for the sample quality issue at $\delta=0$.

In Fig. 2, we plotted the lattice parameters of the #1–3 samples simultaneously, confirming high reproducibility of the sample quality. The high reproducibility however does not secure that the nominal quantity δ is equivalent to the true quantity; we thus measured the net oxygen quantity as

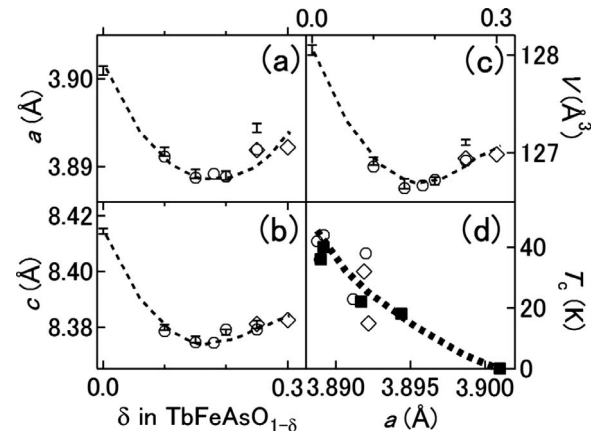


FIG. 2. Unit-cell size evolution of TbFeAsO_{1- δ} through the oxygen quantity; (a–c) show a , c , and the cell volume, respectively. (d) shows magnetic T_c vs a . Open symbols are data obtained from the #2 (circle) and the #3 (diamond) groups of the samples. Dotted curves are a guide to the eye.

follows. We measured the weight of the high-pressure specimen (sample plus Pt tube) before and after the high-pressure run finding no substantial difference. Thus, the net oxygen content of the sample was kept unchanged during the heating. In order to confirm this we analyzed the net oxygen contents of selected samples by a gravimetric method. Nearly 10 mg of a selected sample was fully oxidized at 1450~1500 °C in air or oxygen then was slowly cooled. The final product was found to be TbFeAsO_3 by an XRD method indicating that the reaction of this procedure is $\text{TbFeAsO}_y + (4.5-y)/2\text{O}_2 = \text{TbFeAsO}_3 + 1/2 \text{As}_2\text{O}_3\uparrow$. From the weight loss during the procedure, the net oxygen contents were calculated to be 0.71 and 0.64 for the two selected samples with the nominal oxygen contents of 0.75 ($\delta=0.25$) and 0.70 ($\delta=0.3$), respectively. The experimental oxygen content is slightly smaller than the initial value but the agreement was fairly good.

Let us go back to see the δ dependence of the tetragonal cell dimension of $\text{TbFeAsO}_{1-\delta}$ shown in Figs. 2(a)–2(c). The unit-cell volume decreases with increasing δ and shortly hits the bottom at δ of 0.15–0.18: the change reaches -1.1% almost isotropically (0.3% shrinks along a axis and 0.4% along c axis). The unit-cell volume then turns to increase slightly by further introducing amount of the oxygen vacancies. The overall feature is thus out of the Vegard's law, resulting from a probable competition between the increased mean ionic size of Fe and the unit-cell size reduction due to the oxygen missing. Interestingly, alike features including the broad minimum in the plots was observed for $\text{NdFeAsO}_{1-\delta}$.¹⁸ In large contrast, 0.2% increment in the unit-cell volume was observed for the F-doped Tb system.¹⁹

We examined a relation between the lattice parameters and T_c : T_c vs a is plotted in Fig. 2(d). The T_c was estimated from the magnetic data shown later. It is clear that the lattice-parameter change is correlated with T_c change, as found in $\text{NdFeAsO}_{1-\delta}$ and $\text{NdFeAs}(\text{O},\text{F})$.^{14,20} It is notable that not only the “underdoped” superconductor, but also the “overdoped” superconductor likely follows a unique curve in contrast to what was observed for $\text{NdFeAsO}_{1-\delta}$.²¹

Figures 3(a)–3(e) show temperature and δ dependence of the dc magnetic susceptibility of $\text{TbFeAsO}_{1-\delta}$ measured in a field of 10 Oe. At $\delta=0$, the sample does not show a superconducting drop down to 2 K, while others ($\delta=0.1$ –0.30) show clearly superconducting transitions. The series of the #2 samples show nearly identical properties (shown by solid curves) with those for the same δ #1 samples, confirming reproducibility of the magnetic properties, except at $\delta=0.25$. A rigid shift in the curves along the vertical axis was observed for the $\delta=0.25$ #2 sample probably due to magnetic impurities in the sample accidentally produced.

The highest T_c of 44 K was observed at $\delta=0.18$, and the shielding fraction was also optimized at $\delta=0.18$: it is 1.15 (5 K, ZFC curve), while 1.00 is expected for the perfect diamagnetism. Although superconducting transitions were observed between δ of 0.1 and 0.30, the transitions were relatively broad even at the T_c optimized δ of 0.18. Regarding the copper oxide superconductors, a much sharper transition was commonly observed even for a polycrystalline sample. The broad transition thus suggests that further improvements of the sample quality might be possible.

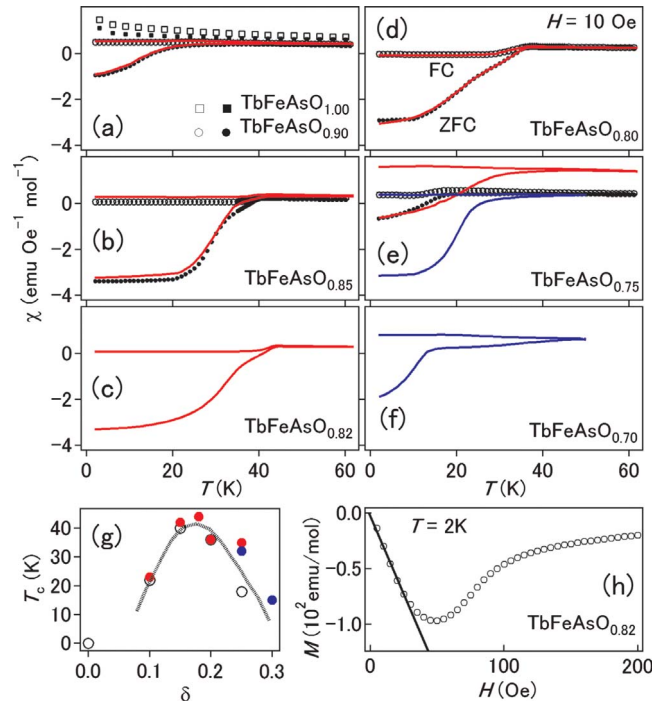


FIG. 3. (Color online) Temperature and oxygen concentration dependence of the dc magnetic susceptibility ($H=10$ Oe) of the polycrystalline $\text{TbFeAsO}_{1-\delta}$, where (a) $\delta=0$, (b) 0.1, (c) 0.15, (d) 0.18, (e) 0.20, (f) 0.25, and (f) 0.30. The solid curves in red and blue are data for the #2 and #3 groups of the samples. (g) Summary of the magnetic T_c vs the oxygen concentration. (h) shows the initial isothermal magnetization (2 K) of the T_c optimized sample $\text{TbFeAsO}_{0.82}$.

The observed T_c is plotted against δ in Fig. 3(g), showing a nearly bell-shaped feature. Compared with the results for $\text{NdFeAsO}_{1-\delta}$,¹³ the highest T_c of 51 K was found at $\delta \sim 0.17$ by a neutron-diffraction method, being comparable with the result for $\text{TbFeAsO}_{1-\delta}$. Although an overdoped feature was expected at $\delta > 0.17$ for $\text{NdFeAsO}_{1-\delta}$, the synthesis condition might not allow synthesizing the overdoped Nd sample, unfortunately.

In order to evaluate the lower critical field (H_{c1}), we measured the initial diamagnetic curve of the highest T_c sample ($\delta=0.18$) at 2 K; the curve is shown in Fig. 3(h), indicating that H_{c1} is approximately 20 Oe, being much lower than that of the other LnFeAsO superconductors.¹³ Meanwhile, it appears that the H_{c1} is nearly comparable with the applied magnetic field in the magnetic-susceptibility measurements, implying a possible origin in part of the relatively broad transitions.

Figures 4(a) and 4(b) show temperature and δ dependence of ρ of the polycrystalline $\text{TbFeAsO}_{1-\delta}$, showing varieties of T_c and normal-state resistivity. The T_c on-set changes in a comparable way with what was found in the magnetic study. The maximum T_c on-set ($\delta=0.18$) estimated from the resistivity data by a graphical analysis is 44 K, being consistent with the magnetic bulk T_c . Meanwhile, the normal-state behavior is rather unusual: for example, the room-temperature ρ goes down with introducing the oxygen vacancies and hits the bottom at the highest T_c composition. The room-

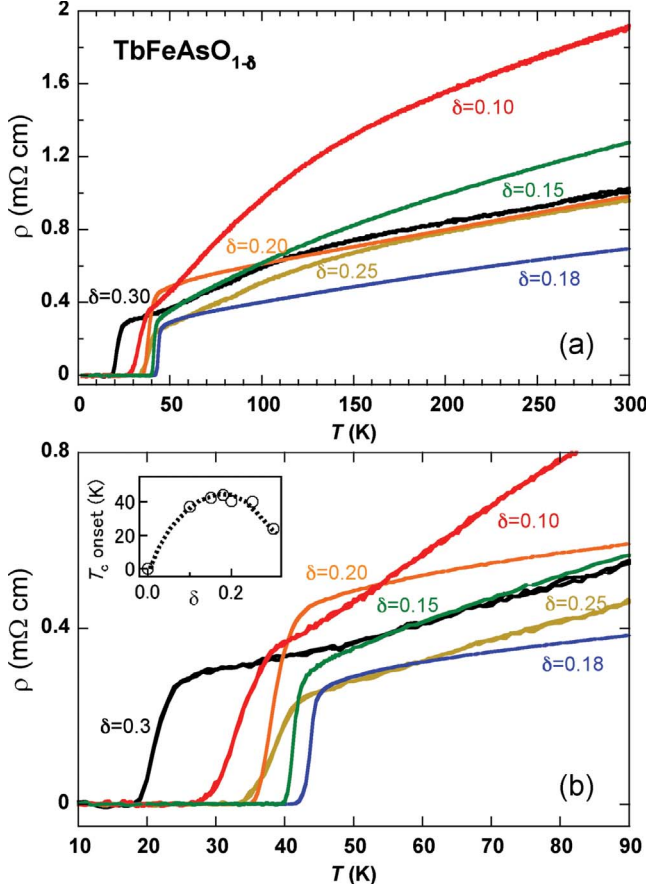


FIG. 4. (Color online) (a) Temperature and oxygen concentration dependence of the electrical resistivity of the polycrystalline $\text{TbFeAsO}_{1-\delta}$, and (b) the expansion of the data. Inset shows T_c onset vs δ .

temperature ρ turns to increase by further accommodation of the oxygen vacancies. As it is widely accepted, charge transport in normal state is highly sensitive to the polycrystalline issue, including degree of sintering, grain boundaries, and surface impurities. Those factors often mask true nature of the charge transport. Thus, quantitative analysis regarding normal-state conductivity is thus left for future work after a high-quality single crystal becomes available.

Figure 5 shows temperature and δ dependence of C_p of the polycrystalline $\text{TbFeAsO}_{1-\delta}$. First, lattice contribution was analyzed by a linear combination of the Debye model and the Einstein model,

$$C(T) = n_D 9 N_A k_B \left(\frac{T}{T_D} \right)^3 \int_0^{T_D/T} \frac{x^4 e^x}{(e^x - 1)^2} dx + n_E 3 N_A k_B \left(\frac{T_E}{T} \right)^2 \frac{e^{T_E/T}}{(e^{T_E/T} - 1)^2},$$

where N_A is the Avogadro's constant, k_B is the Boltzmann's constant, and T_D and T_E are the Debye and the Einstein temperatures, respectively. The scale factors n_D and n_E correspond to the number of vibrating modes per the formula unit in the Debye and the Einstein models, respectively. The fit was carried out between 25 and 80 K for every samples, and

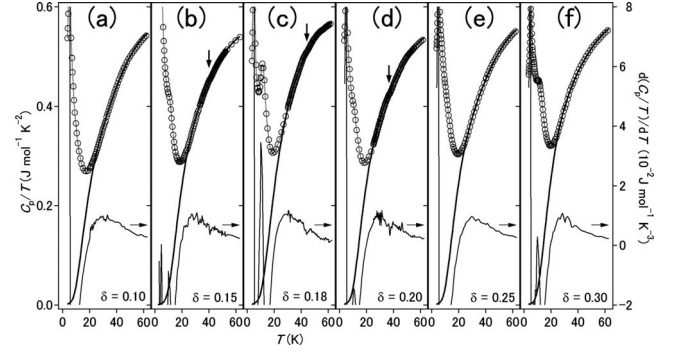


FIG. 5. Oxygen concentration and temperature dependence of the specific heat of the polycrystalline $\text{TbFeAsO}_{1-\delta}$, where (a) $\delta = 0.1$, (b) 0.15, (c) 0.18, (d) 0.2, (e) 0.25, and (f) 0.30. Fat solid curves are fit to the data by using the Debye and Einstein lattice models, and thin solid curves are temperature differential of the data.

the best fit was obtained at T_D of 324(3) K, $T_E = 80.6(5)$ K, $n_D = 0.716(4) \times 3.90$, $n_E = 0.118(5) \times 3.90$ for the $\delta = 0.10$ sample, and T_D of 367(3) K, $T_E = 89.0(7)$ K, $n_D = 0.774(8) \times 3.70$, $n_E = 0.191(5) \times 3.70$ for the $\delta = 0.30$ sample. For the other samples between δ of 0.10 and 0.30, the refined parameter each was found to be intermediate between those, indicating the lattice contribution in C_p is less dependent on the oxygen quantity change. We found T_D for $\text{TbFeAsO}_{1-\delta}$ is almost comparable with T_D for the other superconducting iron compounds.^{22–24}

The fit to the C_p data (fat solid curves in Fig. 6) makes clear that the magnetic contribution at low temperature is substantially large. The sizable contribution probably comes from the localized magnetic moments of Tb^{3+} , which may be less relevant to the observed superconductivity. Detailed analysis of the large magnetic term is thus left for future work.

We carefully searched for an anomaly in the C_p data around T_c , finding a steplike feature (indicated by arrows) at

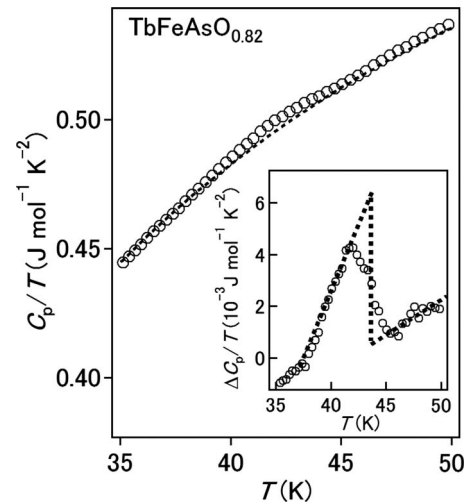


FIG. 6. Expansion of the C_p data for the T_c optimized sample. The fit to the data by the lattice model was shown as a dotted curve and the difference between the data and the fit is shown in the inset.

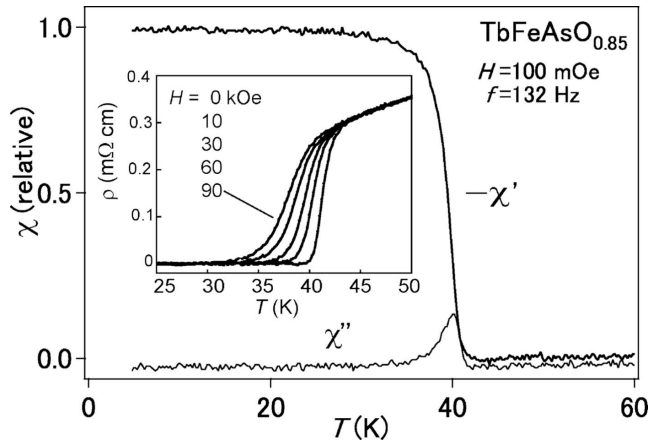


FIG. 7. Temperature dependence of the ac magnetic susceptibility and (inset) temperature and magnetic field dependence of the electrical resistivity of the polycrystalline $\text{TbFeAsO}_{0.85}$.

$\delta=0.15, 0.18, \text{ and } 0.20$. For a convenience, temperature differential curve of the C_p is shown at the bottom of the panel each (thin solid curve), indicating a spike at T_c beyond statistical background. It appears that the steplike anomaly changes as a function of δ in a comparable way with what was observed in the magnetic and the electrical resistivity measurements. At $\delta=0.1, 0.25, \text{ and } 0.30$, expected anomalies are masked by the sizable Tb^{3+} contribution.

Figure 6 shows an expansion of the C_p/T vs T around T_c for the $\delta=0.18$ sample. The dotted curve is the lattice contribution estimated by the analysis using the Debye and the Einstein models as discussed above. The subtracted part is shown in the inset to Fig. 6, showing a superconducting jump. The midpoint of the thermodynamical transition is 44 K, which matches well with the magnetic T_c . $\Delta C_p/T_c$ is estimated to be approximately $5 \text{ mJ mol}^{-1} \text{ K}^{-2}$. A simple Bardeen-Cooper-Schrieffer prediction indicates that the coefficient of the linear term in the specific heat in the normal state (γ) is $\Delta C_p/1.43T_c$ for a weakly coupled superconductor. From the present result, we thus estimated γ of $3.5 \text{ mJ mol}^{-1} \text{ K}^{-2}$ for the $\delta=0.18$ sample. Due to the large Tb^{3+} contribution and the relatively higher T_c , γ is unable to be directly measured for a comparison. However, γ of the related superconductors $\text{LaFe}_{1-x}\text{Co}_x\text{AsO}$ ($x=0.05, 0.11, 0.15$) was directly measured to be approximately $3\text{--}8 \text{ mJ mol}^{-1} \text{ K}^{-2}$,²³ being comparable with the present result for the Tb-1111 system. The F-doped LaFeAsO has also a comparable γ of $3\text{--}8 \text{ mJ mol}^{-1} \text{ K}^{-2}$.²² The estimated γ for the $\delta=0.18$ sample is thus not far from that for the other 1111 superconductors. Besides, the jump $\Delta C_p/T_c$ for the F-doped SmFeAsO was found approximately $6\text{--}8 \text{ mJ mol}^{-1} \text{ K}^{-2}$,²⁵ indicating γ of the superconducting LnFeAsO is much smaller regardless of size of Ln than that of the hole-doped FeAs-122 .

Obviously, the superconducting properties of $\text{TbFeAsO}_{1-\delta}$ are optimized around δ of $0.15\text{--}0.18$. Figure 7 shows ac susceptibility of the $\delta=0.15$ sample. The real part (χ') shows a steep change at T_c , which accords with what was observed in the dc susceptibility measurements. Besides, the imaginary part (χ'') consists of a rather single peak than multiple peaks,

suggesting high degree of chemical homogeneity of the polycrystalline sample. The inset to Fig. 7 shows magnetic field dependence of the electrical resistivity of the $\delta=0.15$ sample. At $H=0$ kOe, the superconducting transition is sharp, even though the sample is polycrystalline. The transition becomes slightly broad with increasing magnitude of the applied magnetic-field. A graphical analysis on the curves indicates that the T_c on-set goes down ~ 2 K in a field of 90 kOe, suggesting high magnitude of the upper critical field (H_{c2}). Unfortunately, the experimental range is too narrow to properly estimate $H_{c2}(T=0$ K), however an attempt using an analytical formula $\mu_0 H_{c2} = -0.693(dH_{c2}/dT)_{T=T_c} T_c$, developed by Werthamer *et al.*,²⁶ indicated that the $\mu_0 H_{c2}(0)$ is near the Pauli limit of 77.5 T ($\mu_0 H_{\text{Pauli}} = 1.24 k_B T_c / \mu_B$, where $T_c = 42$ K). It is thus clear at least that $H_{c2}(0)$ of $\text{TbFeAsO}_{0.85}$ is much lower than what was found for a single-crystal $\text{NdFeAsO}_{0.88}\text{F}_{0.12}$ (300 T for $\mu_0 H_{c2,ab}$).¹¹

Consequently, we studied the atomic structure of the T_c optimized sample ($\delta=0.15$) by a SXR method. The allied sample $\text{LaFeAsO}_{0.85}$ prepared under the same synthesis condition was studied for a comparison. The program RIETAN-2000 was used to analyze the SXR patterns by means of a Rietveld method.²⁷ A neutron-diffraction study for the F-doped LaFeAsO reported elsewhere suggested that the room-temperature structure is well described by a tetragonal model having symmetry $P4/nmm$, while another model having $Cmma$ (orthorhombic) was found below 155 K.²⁸ We employed the tetragonal model as an initial model to parameterize the room-temperature structure of the oxygen vacant $\text{LaFeAsO}_{0.85}$ and $\text{TbFeAsO}_{0.85}$. Coefficients for analytical approximation to the atomic scattering factors were taken from Ref. 29. The pseudo-Voigt function of Toraya was used as a profile function.³⁰ The SXR background was characterized by an 11th-order Legendre polynomial function. Isotropic atomic displacement parameters (B_{iso}) and isotropic Debye-Waller factor, $\exp(-B_{\text{iso}} \sin^2 \theta / \lambda^2)$, were assigned to all the atoms.

First, we studied the structure of $\text{LaFeAsO}_{0.85}$. In a course of the SXR profile analysis, small contributions from impurities, presumably Fe_2O_3 and FeAs , were detected. Since the impurity peaks were too small to estimate those crystallographic parameters correctly, we refined only those scale factors and lattice constants. Final refinements for $\text{LaFeAsO}_{0.85}$ were conducted simultaneously with the impurities peaks. The weight fraction finally estimated was 6.7% for Fe_2O_3 , 0.81% for FeAs , and 92.5% for $\text{LaFeAsO}_{0.85}$. The R factors in total were below 2.3% and the difference curve was found fairly smooth [shown at the bottom of Fig. 8(a)], indicating high-quality of the solution. It appeared that the reasonable solution for $\text{LaFeAsO}_{0.85}$ was indeed attained by using the tetragonal model. At the end of the analysis, we fixed the oxygen thermal parameter to test stability of the oxygen occupancy factor in the analysis. A brief summary of the SXR analysis is in Table I. The result secured that the tetragonal structure model and the analysis method appropriate for the oxygen vacant system prepared under the high-pressure conditions.

We analyzed the SXR profile for the $\delta=0.15$ sample as well. We repeated the same steps applied for $\text{LaFeAsO}_{0.85}$; the refinement steadily reached a reasonable solution. The R

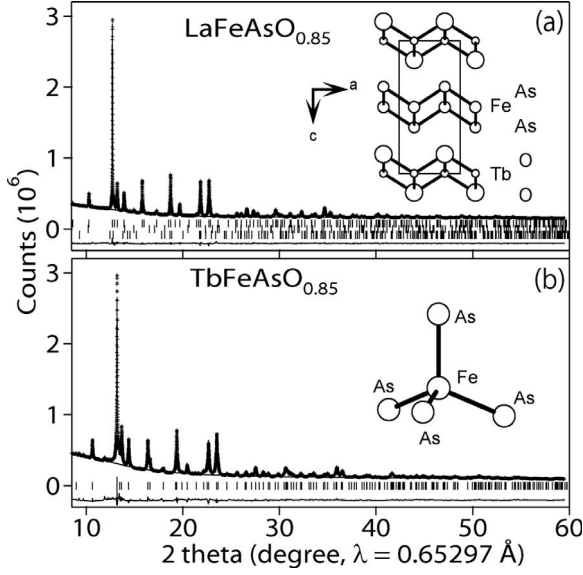


FIG. 8. Rietveld analysis of synchrotron x-ray profiles of (a) $\text{LaFeAsO}_{0.85}$ and (b) $\text{TbFeAsO}_{0.85}$. Cross markers and solid lines show the observed and calculated profiles, respectively. The difference curve is at the bottom each. The positions of Bragg reflections are marked by ticks. Center and bottom lines of ticks are for the impurities Fe_2O_3 and FeAs , respectively. Insets show structural images, based on the present result.

factors were below 2.9%. The impurities Fe_2O_3 and FeAs were not detected above the background, while small amount of other impurities including TbAs (Ref. 15) were found instead. Although the R factors were certainly low, the difference curve especially in the vicinity of the main peak [Fig. 8(b)] suggests further improvement is likely possible. Additional studies including an electron microscopy of the oxygen vacancies and local structure distortion/tilting may assist further improvements.

TABLE I. Structure parameters of $\text{LaFeAsO}_{0.85}$ and $\text{TbFeAsO}_{0.85}$ determined by a synchrotron x-ray powder-diffraction method at room temperature. Space group is $P4/nmm$ (No. 129), $Z=2$, $a=4.02397(4)$ Å, $c=8.71513(8)$ Å, $V=141.118(2)$ Å³, and $\rho_{\text{cal}}=6.664$ g/cm³ for $\text{LaFeAsO}_{0.85}$, and $a=3.89320(4)$ Å, $c=8.38458(14)$ Å, $V=127.085(3)$ Å³, and $\rho_{\text{cal}}=7.887$ g/cm³ for $\text{TbFeAsO}_{0.85}$. R factors were $R_{\text{wp}}=2.15\%$, $R_{\text{p}}=1.48\%$, and $R_{\text{F}}=2.29\%$ for $\text{LaFeAsO}_{0.85}$, and $R_{\text{wp}}=2.83\%$, $R_{\text{p}}=2.13\%$, and $R_{\text{F}}=1.88\%$ for $\text{TbFeAsO}_{0.85}$. Selected bond lengths (d) and angles (θ) are $d_{\text{Fe-As}}=2.4170$ Å($\times 4$), $d_{\text{La-O}}=2.3745$ Å($\times 4$), $\theta_{\text{Fe-As-Fe}}=112.701^\circ$, 72.119° , $\theta_{\text{As-Fe-As}}=107.881$, 112.702° for $\text{LaFeAsO}_{0.85}$, and $d_{\text{Fe-As}}=2.3883$ Å($\times 4$), $d_{\text{Tb-O}}=2.2631$ Å($\times 4$), $\theta_{\text{Fe-As-Fe}}=109.190^\circ$, 70.388° , $\theta_{\text{As-Fe-As}}=109.612$, 109.190° for $\text{TbFeAsO}_{0.85}$.

Site	Wyckoff position	g	x	y	z	B (Å ²)
La	2c	1	0.25	0.25	0.14469(5)	0.557(8)
Fe	2b	1	0.75	0.25	0.5	0.526(18)
As	2c	1	0.25	0.25	0.65367(8)	0.692(13)
O	2a	0.845(5)	0.75	0.25	0	1.5 ^a
Tb	2c	1	0.25	0.25	0.13766(6)	0.709(12)
Fe	2b	1	0.75	0.25	0.5	0.99(3)
As	2c	1	0.25	0.25	0.66502(11)	0.94(2)
O	2a	0.76(1)	0.75	0.25	0	1.5 ^a

^aFixed values.

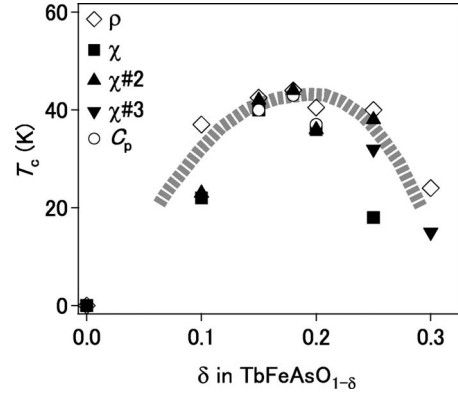


FIG. 9. Summary of T_c change observed for $\text{TbFeAsO}_{1-\delta}$ through varieties of the properties measurements. The marks #2 and #3 indicate data points for the #2 and #3 groups of the samples, respectively.

Insets to Figs. 8(a) and 8(b) show structure images, drawn on the present results for the $\delta=0.15$ sample. Since the coordination environment of Fe was suggested to be coupled with the superconducting properties,¹⁴ we carefully investigated the bond angle between Fe and coordinated As for both $\text{LaFeAsO}_{0.85}$ and $\text{TbFeAsO}_{0.85}$. The calculated bond angles (As-Fe-As) were 113° , 108° for $\text{LaFeAsO}_{0.85}$, and 109° , 110° for $\text{TbFeAsO}_{0.85}$. It appeared that the angles are fairly close to 109.47° for the regular tetrahedron. As discussed in Ref. 14, T_c is probably optimized at the ideal angle. It is thus interesting to investigate possible correlation between the bond angle and amount of oxygen vacancies, which may be coupled with T_c . In order to evaluate the possible correlations further, additional studies are in progress.

In the F-doped LnFeAsO system, a possible accompany of oxygen vacancies with the dopant is usually unexcused and its role remains ambiguous; in short, the true chemical composition should be much closer to $\text{LnFeAsO}_{1-\delta}\text{F}_x$ than

$\text{LnFeAsO}_{1-x}\text{F}_x$.³¹ Studies of the F-free superconductors $\text{LnFeAsO}_{1-\delta}$ are thus expected to exclude additional complexity. In this study, a nearly bell-shaped feature in T_c vs δ (summary is shown in Fig. 9) was observed for the first time to our best knowledge possibly because the complexity is excluded to some extent. Perhaps, the smaller element of Tb may also be help to accept such the relatively large amount of oxygen vacancies in the structure, tuning the superconducting state from the underdoped to the overdoped. The chemical analysis confirmed the true oxygen quantity is close to the nominal.

At the end, the similarity of the bell-shaped feature between the iron and the copper oxide superconductors implies possible common physics. We have thus an open question

that what is the principal origin to reduce the T_c in the overdoped region. Since an overdoped superconductivity is found for the Fe superconductor as well as for the Cu superconductor, continuous efforts are directed to investigate nature of the superconductivity of the overdoped $\text{TbFeAsO}_{1-\delta}$.³²

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